Name of the Digitization of chemistry experiments to improve the quality and

project: support chemistry teaching in secondary schools

Acronym: ChemIQSoc

Project 2021-1-SK01-KA220-VET-000027995

number:



Tittle: Preparation of paracetamol

Work instructions

Task: Prepare 4-hydroxyacetanilide (paracetamol) by acetylation of 4-aminophenol.

Theory

Acetylation of 4-aminophenol is the addition-elimination reaction of an aromatic amine with acetic anhydride to form paracetamol according to Eq:

$$HO \longrightarrow NH_2 \longrightarrow HO \longrightarrow NHCOCH_3$$

Equipment: boiling flask (100 ml), magnetic stirrer with stir bar, graduated cylinder, reflux condenser, cooling water inlet and outlet hose, Büchner funnel, rubber adapter, suction flask, water jet pump, funnel, filter paper, beaker, laboratory stand

Chemicals: 4-aminophenol, acetic anhydride

Procedures:

1. All work is carried out with safety goggles!

Acetylation of 4-aminophenol

- 1. Mix 2.1 g of 4-aminophenol with 18 ml of water in a 100 ml flask with a stir bar and add 3 ml of acetic anhydride to the suspension. Place a reflux condenser on the flask and reflux the reaction mixture for 15 min with stirring.
- 2. Cool the flask in ice-water, aspirate the solid that has fallen out on a Büchner funnel, wash with cold water and dry the raw product by suction.

Crystallisation of paracetamol

1. Add a small quantity of water to the raw paracetamol in the boiling flask, fit a reflux condenser and bring the suspension to the boil. In small batches, add water through the condenser until a saturated boiling solution is formed. When hot, filter it through a folded paper filter. Cool the filtrate, aspirate the precipitated crystals on a Büchner funnel and dry.

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Management of chemical substances

Chemicals	Form	H-statements	P-statements
4-Aminophenol	Solid, raw	1	P261, P273, P280, P301 + P310, P311
Acetic anhydride	Liquid, 99%	H226, H302, H314, H332	P280, P305 + P351 + P338, P310

Sources of risk and assessment of risk severity

There is no risk when following all the principles for working with chemicals and using personal protective equipment (gloves, goggles, lab coat).

Waste management method

Dispose of waste materials in a marked container. Do not return unconsumed residues to storage bottles. Dispose of broken glass in a marked container.

Risk reduction measures

Avoid direct exposure, use protective equipment. Do not expose to prolonged or repeated exposure. In the event of an accident or if you feel unwell, inform the teacher immediately. These substances or their containers must be disposed of as hazardous waste. Do not eat, drink, smoke. Wash your hands with warm water and soap or treat them with a reparative cream after work or when interrupting work.